Condensed Thiophen Ring Systems. Part X.¹ Synthesis and Reactions of 2-Aryl-1H-[1]benzothieno[2,3-b]pyrroles and 2-Aryl-1H-[1]benzothieno[3,2-b]pyrroles²

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Reductive cyclisation of a trans-1-aryl-2-(3-nitro-2-benzo[b]thienyl)- or a trans-1-aryl-2-(2-nitro-3-benzo[b]thienyl)-ethene, respectively, with triethyl phosphite gave the title compounds. The trans-alkenes were prepared by the Wittig reaction from 3-nitrobenzo[b]thiophen-2-carbaldehyde or 2-nitrobenzo[b]thiophen-3-carbaldehyde, respectively. Improved methods for the preparation of these aldehydes are reported, together with some electrophilic substitution reactions of 2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole.

REDUCTIVE cyclisation of cis- or trans-o-nitrostilbene or α -nitrostilbene with triethyl phosphite yields 2-phenylindole.^{3,4} β-Substituted o-nitrostyrenes similarly give the corresponding 2-substituted indoles.3-6 Indoles may be prepared also by thermal decompositions of o-6,7 or β -azidostyrenes,^{8,9} which proceed via the intermediacy of a nitrene.⁶ Cyclisation of the nitro-compounds may or may not involve a nitrene intermediate.3-6, 10, 11

We now report that the 2-aryl-1H-[1]benzothieno-[3,2-b] pyrroles (1; R = H, Me, or Cl) and the 2-aryl-1H-[1] benzothieno[2,3-b] pyrroles (2; R = H, Me, or

¹ Part IX, K. E. Chippendale, B. Iddon, and H. Suschitzky,

J.C.S. Perkin I, 1972, 2030. ² Presented at the Annual Meeting of The Chemical Society (Heterocyclic Section), Manchester, 10-14 April, 1972, abstract no. 9.9.

³ J. I. G. Cadogan, *Quart. Rev.*, **1968**, **22**, 222. ⁴ J. I. G. Cadogan, M. Cameron-Wood, R. K. Mackie, and R. J. Sundbarg J. Over Ch. 1965, 4831.

R. J. Sundberg, J. Org. Chem., 1965, 30, 3604.

Cl) may be prepared in yields of 65-80% by reductive cyclisation of the 1-aryl-2-(3-nitro-2-benzo[b]thienyl)ethenes (3; R = H, Me, or Cl) or the 1-aryl-2-(2-nitro-3-benzo[b]thienyl)ethenes (4; R = H, Me, or Cl), respectively, with a large excess ⁵ of triethyl phosphite. The 3-proton of the benzothienopyrroles (1) and (2) gave rise to a doublet in the n.m.r. spectrum at $\tau 2.95$ — 3.25 ($J_{1.3}$ 1.5—2.0 Hz), collapsing to a singlet on Ndeuteriation.

1H-[1]Benzothieno[3,2-b]pyrrole has been prepared

⁶ R. J. Sundberg, H. F. Russell, W. V. Ligon, and Long-Su Lin, J. Org. Chem., 1972, **37**, 719. ⁷ R. J. Sundberg, Long-Su Lin, and D. E. Blackburn, J. Heterocyclic Chem., 1969, **6**, 441.

⁸ G. Smolinsky and C. A. Pryde, J. Org. Chem., 1968, 33, 2411.

⁹ K. Isomura, S. Kobayashi, and H. Taniguchi, Tetrahedron Letters, 1968, 3499.

¹⁰ R. J. Sundberg, J. Amer. Chem. Soc., 1966, 88, 3781.

¹¹ J. I. G. Cadogan and A. Cooper, J. Chem. Soc., (B) 1969, 883.

by reductive cyclisation of 3-nitro-2-(β -nitrovinyl)benzo[b]thiophen,^{12,13} and by reductive cyclisation of β -(3-nitro-2-benzo[b]thienyl)pyruvic acid followed by decarboxylation of the resulting 2-carboxylic acid.¹³ It is substituted by electrophiles in the 2-position ^{14,15} as predicted by molecular orbital calculations.¹⁵ As far as we are aware, compound (5) is the only derivative of 1*H*-[1]benzothieno[2,3-b]pyrrole reported. It may be prepared by treatment of the tin double salt of 3-aminobenzo[b]thiophen with methyl 1-hydroxy-1methylethyl ketone in ethanol in the presence of anhydrous zinc chloride.¹⁶ Compound (6) may be prepared similarly.¹⁶

Successive treatment of an ethereal solution of benzyltriphenylphosphonium chloride with n-butyllithium and 3-nitrobenzo[b]thiophen-2-carbaldehvde gave trans-1-(3-nitro-2-benzo[b]thienyl)-2-phenylethene (3; R = H) (27%). The yield of this alkene was increased to 68% by using tetrahydrofuran as solvent. The trans-alkenes (3; R = Me or Cl) and (4; R = H, Me, or Cl) were prepared similarly in tetrahydrofuran (50-70% yields). We did not isolate any of the cisisomers of compounds (3) and (4) (R = H, Me, or Cl), presumably because formation of these compounds is sterically hindered by the presence of a bulky nitrogroup in the nitro-aldehyde starting materials. The trans stereochemistry of compounds (3) and (4) was indicated by strong i.r. absorption near 960 cm⁻¹ and by a coupling of 16.0 Hz between the two vinyl protons (cf. ref. 5).

3-Nitrobenzo[b]thiophen-2-carbaldehyde was prepared by acidic hydrolysis of the anil (7). The latter was obtained by treatment of 2-benzo[b]thienyl-lithium with dimethyl sulphate followed by nitration and condensation of the resulting 2-methyl-3-nitrobenzo[b]thiophen with p-nitroso-NN-dimethylaniline. Our procedure allows the aldehyde to be prepared in 57% overall yield from benzo[b]thiophen whereas that described by Shkurko and Mamaev¹⁷ gives the aldehyde in only 15% yield from 2-hydroxymethylbenzo[b]thiophen. Nitration of benzo[b]thiophen-2-carbaldehyde gives a mixture of isomers.¹⁸

2-Nitrobenzo[b]thiophen-3-carbaldehyde was prepared similarly from benzo[b]thiophen via 3-methylbenzo[b]thiophen. The low overall yield (9%) in this case can be attributed in part to the fact that the methyl group in 3-methyl-2-nitrobenzo[b]thiophen is less active towards condensation with p-nitroso-NN-dimethylaniline than that in its 2-methyl-3-nitro-isomer, and in part to the low yield of 3-methyl-2-nitrobenzo[b]thiophen ob-

O. P. Shkurko and V. P. Mamaev, *Khim. geterotsikl. Soedinenii*, 1966, 634 (*Chem. Abs.*, 1967, 66, 65,410).
O. P. Shkurko and V. P. Mamaev, *Izvest. sibirsk. Otdel.*

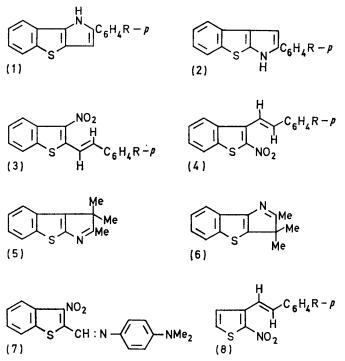
¹³ O. P. Shkurko and V. P. Mamaev, *Izvest. sibirsk. Otdel. Akad. Nauk. S.S.S.R., Ser. khim. Nauk,* 1967, 112 (Chem. Abs., 1968, **69**, 27,290).

¹⁴ O. P. Shkurko and V. P. Mamaev, *Izvest. sibirsk. Otdel. Akad. Nauk S.S.S.R., Ser. khim. Nauk*, 1967, 98 (Chem. Abs., 1969, **70**, 3876, 47,330).

¹⁵ O. P. Shkurko, *Izvest. Akad. Nauk S.S.S.R., Ser. khim.*,
1968, 184 (*Chem. Abs.*, 1968, **69**, 26,664).
¹⁶ V. G. Zhiryakov and P. I. Abramenko, U.S.S.R. Pat.

¹⁶ V. G. Zhiryakov and P. I. Abramenko, U.S.S.R. Pat. 166,700/1964 (*Chem. Abs.*, 1965, **62**, 10,438).

tained by nitration of 3-methylbenzo[b]thiophen.¹⁹ Nitration of benzo[b]thiophen-3-carbaldehyde with fuming nitric acid in a mixture of acetic acid and acetic anhydride is reported ²⁰ to give 2-nitrobenzo[b]thiophen-3-carbaldehyde, but using this method we obtained only a mixture of nitro-isomers in agreement with others.²¹



Since the completion of our work, Srinivasan *et al.*²² have reported that 2-nitro-3-methylthiophen condenses with benzaldehyde and *p*-anisaldehyde to give the *trans*-alkenes (8; R = H or OMe, respectively). An extension of this reaction to the present work would possibly allow the *trans*-alkenes (4) to be prepared more conveniently.

Since 2-phenyl-1*H*-[1]benzothieno[3,2-*b*]pyrrole (1; R = H) was the most accessible compound, we studied its chemistry briefly. This compound is unstable in the presence of strong acids. Its attempted bromination with bromine in chloroform gave only tar. Successive treatment with n-butyl-lithium and dimethyl sulphate gave a high yield of 1-methyl-2-phenyl[1]benzothieno[3,2-*b*]pyrrole, which gave 3-bromo-1-methyl-2-phenyl[1]benzothieno[3,2-*b*]pyrrole on bromination with bromine in chloroform. Reaction of the bromo-

¹⁷ O. P. Shkurko and V. P. Mamaev, *Izvest. sibirsk. Otdel. Akad. Nauk S.S.S.R., Ser. khim. Nauk*, 1965, 81 (Chem. Abs., 1966, **64**, 2040).

¹⁸ V. P. Mamaev and O. P. Shkurko, *Khim. geterotsikl. soedin*enii, Akad. Nauk Latv. S.S.S.R., 1965, 516 (Chem. Abs., 1966, **64**, 675).

¹⁹ D. A. Shirley, M. J. Danzig, and F. C. Canter, J. Amer. Chem. Soc., 1953, **75**, 3278.

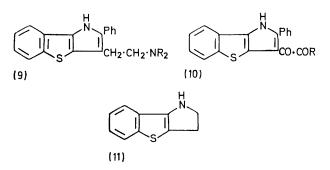
²⁰ Ng. Ph. Buu-Hoi and Ng. Hoan, J. Chem. Soc., 1951, 251.
²¹ G. C. Brophy, S. Sternhell, N. M. D. Brown, I. Brown, K. J. Armstrong, and M. Martin-Smith, J. Chem. Soc. (C), 1970, 933.

²² K. Srinivasan, K. K. Balasubramanian, and S. Swaminathan, *Chem. and Ind.*, 1971, 398.

derivative with n-butyl-lithium followed by carboxylation of the product gave 1-methyl-2-phenyl-[1]benzothieno[3,2-b]pyrrole-3-carboxylic acid.

Vilsmeier–Haack formylation of the 2-phenyl derivative (1; R = H) gave 2-phenyl-1*H*-[1]benzothieno[3,2-*b*]pyrrole-3-carbaldehyde, which was reduced by sodium borohydride to the 3-hydroxymethyl compound. Treatment of this compound with thionyl chloride gave only tar.

The tryptamine analogues (9; R = Me or Et) were prepared through treatment of the pyrrole (1; R = H) with oxalyl chloride in ether followed by treatment of the glyoxyloyl chloride (10; R = Cl) with dimethylamine or diethylamine and reduction of the resulting glyoxylamide (10; $R = NMe_2$ or NEt_2) with lithium aluminium hydride in ether.



In an attempt to prepare the 2,3-dihydro-compound (11), we heated a mixture of 2-ethyl-3-nitrobenzo[b]thiophen and an excess of triethyl phosphite under reflux for 7 h. This gave mainly starting material (70% recovery), together with tar. Treatment of 2-ethyl-3-nitrobenzo[b]thiophen with triethyl phosphite in refluxing xylene or trimethyl phosphate similarly gave mainly starting material.

EXPERIMENTAL

Molecular weights were determined by mass spectrometry. The spectroscopic instruments used in this work and the experimental precautions taken were the same as described previously.²³ N.m.r. signals were singlets unless stated otherwise. Light petroleum refers to the fraction of b.p. 60-80°.

2-Methyl-3-nitrobenzo[b]thiophen (58%), m.p. 96-99° (from ethanol) (lit.,¹⁹ 98-98.5°; lit.,²⁴ 97-98°), and 3-methyl-2-nitrobenzo[b]thiophen (26%), m.p. 146-148° (from ethanol) (lit.,¹⁹ 148-149°), were prepared by literature procedures. For the synthesis of the last compound, 3-methylbenzo[b]thiophen was prepared by our procedure.25

2-Ethyl-3-nitrobenzo[b]thiophen.—Concentrated nitric acid (12 ml) was added during 5 min to a solution of 2-ethylbenzo[b]thiophen ²⁶ (4.0 g, 24.7 mmol) in acetic acid (26 ml) at 0° . The temperature of the mixture was kept at 5-10° during the addition. The mixture was stirred at 5-10° for a further 30 min and then poured on to ice. Extraction with chloroform gave 2-ethyl-3-nitrobenzo[b]thiophen (2.4 g,

²³ Part VIII, K. E. Chippendale, B. Iddon, and H. Suschitzky, J.C.S. Perkin I, 1972, 2023. ²⁴ J. Cooper and R. M. Scrowston, J.C.S. Perkin I, 1972, 265.

48%), m.p. 103–105° (from ethanol), τ (CDCl₃) 1.55–1.68 (m, 4-H), 2.15—2.70 (m, aromatic), 6.62 (q, J 8.0 Hz, CH₂), and 8.59 (t, J 8.0 Hz, CH₃) (Found: C, 58.3; H, 4.4; N, 6.8; S, 15.5%; M, 207. $C_{10}H_9NO_2S$ requires C, 57.95; H, 4.4; N, 6.7; S, 15.5%; M, 207).

3-Nitrobenzo[b]thiophen-2-carbaldehyde.—(i) A mixture of 2-methyl-3-nitrobenzo[b]thiophen (5.0 g, 25.9 mmol), p-nitroso-NN-dimethylaniline (5.8 g, 38.5 mmol), and ethanol (25 ml) was heated under reflux for 5 h. It was then cooled and the precipitate was filtered off, washed with hot ethanol (25 ml), and recrystallised from light petroleum, to give p-dimethylamino-N-(3-nitrobenzo[b]thiophen-2-ylidene)aniline (6.9 g, 82%), m.p. 230-233° (Found: C, 62.3; H, 4.6; N, 12.7%; M, 325. C₁₇H₁₅N₃-O₂S requires C, 62.75; H, 4.65; N, 12.9%; M, 325).

(ii) A mixture of the anil (6.0 g, 18.45 mmol) and concentrated hydrochloric acid (100 ml) was heated under reflux for 1 h. It was then cooled; the precipitate was filtered off, washed with 2n-hydrochloric acid, dried, and recrystallised from benzene-n-hexane (1:1), to give 3-nitrobenzo[b]thiophen-2-carbaldehyde (3.5 g, 92%), m.p. 125—127° (lit., ¹⁹ 125—127°), ν_{max} (Nujol) 1680 cm⁻¹ (C:O), τ (CDCl₃) 1·40 (q, J_o 9·0, J_m 3·0 Hz, 4-H), 1·85— 2.40 (m, aromatic), and -0.75 (CHO).

2-Nitrobenzo[b]thiophen-3-carbaldehyde (77%), prepared similarly, had m.p. 122-124° (sublimed, 120° at 10 mmHg) (lit.,²⁰ 124°), ν_{max} (Nujol) 1680 cm⁻¹ (C:O), τ (CDCl₃) 1·25 (q, J_o 7·5, J_m 2·0 Hz, 4-H), 2·00–2·75 (m, aromatic), and -0·91 (CHO) (Found: C, 52·1; H, 2·5; N, 6·5%; M, 207. C₉H₅NO₃S requires C, 52.2; H, 2.4; N, 6.7; M. 207).

p-Dimethylamino-N-(2-nitrobenzo[b]thiophen-3-ylidene)aniline (59%) had m.p. 165—167° (from ethanol), τ (CDCl₃) 0.33 (CH), 0.5-0.85 (m, 4-H), 2.00-2.68 (m, aromatic), 3.04-3.42 (m, aromatic protons adjacent to p-NMe₂), and 6.92 (Me) (Found: C, 62.6; H, 4.8; N, 13.0%; M, 325).

trans-1-Aryl-2-(3-nitro-2-benzo[b]thienyl)- and trans-1-Aryl-2-(2-nitro-3-benzo[b]thienyl)-ethenes.—General method. n-Butyl-lithium (1.0 equiv.) was added dropwise to a stirred suspension of a benzyltriphenylphosphonium chloride (1.0 equiv.) in tetrahydrofuran (10 ml g⁻¹) at room temperature and the mixture was stirred for a further 15 min. A solution of the nitrobenzo b this phencarbaldehyde (1.1 equiv.) in tetrahydrofuran (10% w/v) was added dropwise and the mixture was heated under reflux for 2 h. It was then cooled, washed with water, dried $(MgSO_{4})$, and chromatographed on an alumina column. Ether-light petroleum (1:1) eluted the product. Details of the products are given in Table 1.

2-Aryl-1H-[1]benzothieno[3,2-b]pyrroles and 2-Aryl-1H-[1]benzothieno[2,3-b]pyrroles.—General method. A mixture of the trans-alkene (Table 1) (1.0 equiv.) and triethyl phosphite (12.0 equiv.) was heated under reflux for 20 h under nitrogen. The excess of triethyl phosphite and the triethyl phosphate produced were then distilled off at 0.1mmHg, and the residue was chromatographed on alumina. The eluant and details of the products are given in Table 2.

1-Methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole.—A solution of n-butyl-lithium (8.05 mmol) in hexane (3.5 ml)was added dropwise to a stirred solution of 2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole (2.0 g, 8.03 mmol) in ether

25 R. P. Dickinson and B. Iddon, J. Chem. Soc. (C), 1968, 2733.

²⁶ E. N. Karaulova, D. Sh. Mellanova, and G. D. Gal'pern, J. Gen. Chem. U.S.S.R., 1960, 30, 3262.

(40 ml) at room temperature, and the mixture was stirred for 15 min. Dimethyl sulphate (1·1 g, 8·7 mmol) in ether (10 ml) was added dropwise during 10 min and the mixture was stirred for a further 1 h. An excess of sodium ethoxide in ethanol was added and the mixture was heated under reflux for 1 h to destroy the excess of dimethyl sulphate. Work-up in the usual way gave the *product* (1·64 g, 78%), m.p. 119—121°, τ (CDCl₃) 1·90—2·73 (m, aromatic), 3·41 (3-H), and 5·92 (Me) (Found: C, 77·3; H, 5·1; N, 5·2%; M, 263. C₁₇H₁₃NS requires C, 77·5; H, 5·0; N, 5·3%; M, 263).

Bromination of 1-Methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole.—Bromine (0.80 g, 5.0 mmol) in chloroform (8.5 ml) was added dropwise during 10 min to a stirred solution of 1-methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole (1.2 g, 4.6 mmol) in chloroform (24 ml) at room temperature and the mixture was stirred at room temperature for a further 2 h. 4.5%; *M*, 307. C₁₈H₁₃NO₂S requires C, 70.3; H, 4.3; N, 4.6%; *M*, 307).

Vilsmeier-Haack Formylation of 2-Phenyl-1H-[1]benzothieno[3,2-b]pyrrole.—The benzothienopyrrole (2.0 g, 8.03 mmol) was added in small portions to a mixture of NN-dimethylformamide (4.3 g, 59.0 mmol), and phosphoryl chloride (1.6 g, 10.4 mmol) at 0°. The mixture was heated at 55° for 90 min, then cooled and poured on to ice. The resulting mixture was made alkaline with saturated aqueous potassium carbonate and the precipitate was filtered off and crystallised from ethanol, to give 2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3-carbaldehyde (1.8 g, 80%), m.p. 248—250°, v_{max} . (Nujol) 3200s (NH) and 1620s cm⁻¹ (C:O), τ (CDCl₃) -0.15 (CHO), 1.78—2.80 (m, aromatic), and 7.38 (exchangeable, NH) (Found: C, 73.6; H, 4.0; N, 5.05%; M, 277. C₁₇H₁₁NOS requires C, 73.6; H, 4.0; N, 5.05%; M, 277).

TABLE]

trans-1-Aryl-2-(3-nitro-2-benzo[b]thienyl)ethenes and trans-1-aryl-2-(2-nitro-3-benzo[b]thienyl)ethenes

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		Yield	νmax. (CH:CH) b	Chemical shifts (τ) •					Found (%)				Required (%)		
Compound	M.p.ø (°C)	(%)	(cm^{-1})	Olefinic H d	4-H	Me	Other	ີດ	н	N	Formula	ć	н	N	
(3; $R = H$)	105-109 •	68	965	1.75(d,	1•55(m)		2·10-2·81(m)	68 •0	4.1	$5 \cdot 0$	$C_{16}H_{11}NO_2S$	68.3	3.95	5.0	
(3; R = Me)	129-132	63	955	J 16.0 Hz) 1.80(d, J 16.0 Hz)	1•59(m)	7.66	2·15-3·00(m)	69·1	4.6	4.7	$\mathrm{C_{17}H_{13}NO_2S}$	69·1	4 ·4	4 ·7	
(3; $R = Cl$)	108111	72	950	j 10 0 112,	1•52(m)		1.85-2.90(m)	60.8	3.2	4.3	$C_{16}H_{10}CINO_2S$	60.85	$3 \cdot 2$	4 · 4 5	
(4; $R = H$) (4; $R = Me$) (4; $R = Cl$)	143144 120125 ¢ 147151 ¢	59 50 61	980 965 970			7.60	2·10-3·00(m) 2·10-2·90(m) 2·15-2·90(m)	68·3 68·9 60·9	4·1 4·5 3·5	4·8 4·7 4·3	C ₁₆ H ₁₁ NO ₂ S C ₁₇ H ₁₃ NO ₂ S C ₁₆ H ₁₆ CINO ₂ S				

• Recrystallised from ethanol except for compound (3; R = H) which was recrystallised from methanol. b Nujol. • Solvent CDCl₃. • In most cases the signals overlapped with those of the aromatic protons and only in the two instances shown was it possible to record the coupling constants. • With decomposition.

TABLE 2

2-Aryl-1H-[1]benzothieno[3,2-b]pyrroles and 2-aryl-1H-[1]benzothieno[2,3-b]pyrroles

		Crvst.		Yield	νmax.(NH) δ	Chemical shifts (τ) \bullet Found (%)								Required (%)			
Compound	M.p. (°C)	from a	Eluant ª	(%)	(cm^{-1})	3-н	Ме	Other	$J_{1,3}/\text{Hz}$	c	Н	N	Formula	С	н	Ň	
(1; R = H)	188—190°	A	A	76	3440	3.00(d)		1.70-2.73(m)	2.0	77.1	4.6	5.4	C ₁₆ H ₁₁ NS	77.1	4.45	5.6	
(1; $R = Me$) (1; $R = Cl$)	178—180 195—197	B B	D E	79 66	3430 3430	3·23(d) 3·22(d)	7.67	$2 \cdot 10 - 2 \cdot 90(m)$ $1 \cdot 85 - 2 \cdot 80(m)$	2·0 2·0	77•5 68•05	5·0 3·7	5·2 4·8	C ₁₇ H ₁₃ NS C ₁₆ H ₁₀ CINS	77·5 67·8	5·0 3·5	5·3 4·9	
(2; R = H)	171-173	в	Ē	78	3440	2.95(d)		1.90-2.73(m)	2.0	77.5	4.8	5.4	$C_{14}H_{11}NS$				
(2; $R = Me$) (2; $R = Cl$)	$166-168 \\ 182-184$	C C	DE	67 64	3450 3420	3·25(d) 2·01(d)	7.62	$2 \cdot 10 - 3 \cdot 10$ (m) $2 \cdot 00 - 2 \cdot 80$ (m)	2·0 1·5	77.5 67.6	4·8 3·7	5·2 4·8	$C_{17}H_{13}NS$ $C_{16}H_{10}CINS$				

• A = Ether, B = benzene-light petroleum (1:1), C = toluene-light petroleum (1:1), D = ether-light petroleum (1:2), E = ether-light petroleum (1:1). • Nujol. • Solvent CDCl₂.

2N-Sodium hydroxide (50 ml) was added and the organic layer was separated, washed with water, and dried (MgSO₄). Removal of the solvent by distillation gave a residue which was chromatographed on alumina. Light petroleum (b.p. 40-60°) eluted 3-bromo-1-methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole (1.4 g, 89%), m.p. 83-85° (from benzene-light petroleum) [Found: C, 59.9; H, 3.3; N, 3.95%; M, 342; M + 2, 344 (ratio 1:1). C₁₇H₁₂BrNS requires C, 59.65; H, 3.5; N, 4.1%; M, 342; M + 2, 344 (ratio 1:1)].

1-Methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole-3-carboxy-

lic Acid.—A solution of n-butyl-lithium (4.6 mmol) in hexane (2.0 ml) was added dropwise to a stirred solution of 3-bromo-1-methyl-2-phenyl[1]benzothieno[3,2-b]pyrrole (1.4 g, 4.1 mmol) in ether (30 ml) at -70° , and the mixture was stirred at -70° for 1 h. An excess of dry, solid carbon dioxide was then added in small pieces and the mixture was allowed to warm slowly to room temperature. After the addition of 2N-hydrochloric acid, the mixture was worked up in the usual way to give the acid (0.5 g, 40%), m.p. 200—202° (from benzene), v_{max} (Nujol) 3100br (OH) and 1670s cm⁻¹ (C:O), τ (CDCl₃) 1.80—2.69 (m, aromatic and OH) and 6.11 (Me) (Found: C, 69.9; H, 4.3; N, 3-Hydroxymethyl-2-phenyl-1H-[1]benzothieno[3,2-b]pyr-

role.—A mixture of 2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3-carbaldehyde (3.0 g, 10.8 mmol), sodium borohydride (0.1 g, 2.6 mmol), ethanol (30 ml), and 4N-sodium hydroxide (10 ml) was heated at 50° for 30 min. It was then cooled and filtered, and an excess of water was added. Extraction with ether gave the *product* (1.0 g, 30%), b.p. 120—125° at 0.2 mmHg (Kugelrohr apparatus), v_{max} (film) 3420br cm⁻¹ (OH) (Found: C, 73.5; H, 4.7; N, 5.0%; M, 279). C₁₇H₁₃NOS requires C, 73.1; H, 4.7; N, 5.0%; M, 279).

Reaction of 2-Phenyl-1H-[1]benzothieno-[3,2-b]pyrrole with Oxalyl Chloride.—A mixture of benzothienopyrrole (1.0 g, 4.02 mmol) and oxalyl chloride (0.67 g, 5.3 mmol) in ether (10 ml) was stirred at room temperature for 20 h. The precipitate was filtered off and dried in air to give 2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3-glyoxyloyl chloride (1.37 g, 100%), m.p. 132—134°, v_{max} . (Nujol) 3280s (NH) and 1600 cm⁻¹ (C:O) (Found: C, 63.3; H, 3.1; N, 3.9. C₁₈H₁₀ClNO₂S requires C, 63.6; H, 3.0; N, 4.1%).

NN-Dimethyl-2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3-glyoxylamide.—A solution of dimethylamine in ethanol (33%; 28 ml) was added to a solution of the glyoxyloyl chloride (7.0 g, 20.6 mmol) in ethanol (28 ml) and the mixture was stirred for 2 h at room temperature. It was then filtered; evaporation of the filtrate to dryness under reduced pressure gave the *product* (6.5 g, 93%), m.p. 210—211° (from ethanol), v_{max} (Nujol) 3240s (NH), and 1650s and 1610s cm⁻¹ (C:O), τ (CDCl₃) 1·90—2·81 (m, aromatic), 7·26 and 7·42 (Me), and 7·49 (exchangeable, NH) (Found: C, 68·9; H, 4·75; N, 8·0%; *M*, 348. C₂₀H₁₆N₂-O₂S requires C, 68·9; H, 4·6; N, 8·0%; *M*, 348).

NN-Diethyl-2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3glyoxylamide (72%), prepared similarly in benzene, had m.p. 218—220° (from ethanol), v_{max} (Nujol) 3180s (NH), and 1630s and 1615s cm⁻¹ (C:O), τ (CDCl₃) 1·92—2·90 (m, aromatic), 6·78 (q, J 8·0 Hz, CH₂), 8·95 (t, J 8·0 Hz, CH₃), and 8·65 (exchangeable, NH) (Found: C, 69·95; H, 5·2; N, 7·4%; M, 376. C₂₂H₂₀N₂O₂S requires C, 70·2; H, 5·35; N, 7·4%; M, 376).

3-(2-Dimethylaminoethyl)-2-phenyl-1H-[1]benzothieno-[3,2-b]pyrrole.—A solution of NN-dimethyl-2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole-3-glyoxylamide (5.0 g, 14.4 mmol) in dry ether (100 ml) was added quickly to a stirred suspension of lithium aluminium hydride (0.2 g, 5.70 mmol) in dry ether (50 ml) under nitrogen at room temperature, and the mixture was heated under reflux for 3 h. Ethyl acetate (10 ml) was added cautiously, followed by 5Nsodium hydroxide (15 ml), and the mixture was then heated under reflux for a further 1 h. The precipitate was filtered off, the residue was washed thoroughly with ethyl acetate, and the filtrate and washings were combined and dried (MgSO₄). Removal of the solvents by distillation under reduced pressure gave a residue which was crystallised several times from light petroleum to give the *product* (4·2 g, 91%), m.p. 109—111°, τ (CDCl₃) 1·29 (exchangeable, NH), 2·10—2·90 (m, aromatic), 6·82—7·52 (m, CH₂), and 7·76 (Me) (Found: C, 74·8; H, 6·5; N, 8·6%; *M*, 320. C₂₀H₂₀N₂S requires C, 75·0; H, 6·3; N, 8·75%; *M*, 320).

3-(2-Diethylaminoethyl)-2-phenyl-1H-[1]benzothieno[3,2-b]pyrrole (76%), prepared similarly, had m.p. 105—107° (from light petroleum), τ (CDCl₃) 2·01—3·00 (m, aromatic), 6·80 (exchangeable, NH), 7·16 (CH₂), 7·42 (q, J 7·5 Hz, CH₂), and 9·00 (t, J 7·5 Hz, CH₃) (Found: C, 75·5; H, 7·0; N, 7·9%; M, 348. C₂₂H₂₄N₂S requires C, 75·8; H, 6·9; N, 8·0%; M, 348).

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